## Telangana State Council of Higher Education, Govt. of Telangana B.Sc., CBCS Common Core Syllabi for all Universities in Telangana (wef 2016-'17)

## PROPOSED SCHEME FOR CHOICE BASED CREDIT SYSTEM IN B.Sc., BIOCHEMISTRY

FIRST YEAR – SEMESTER I						
Code	Course Title	Course Type	HPW	Credits		
BS 101	Environmental Studies	AECC 1	2	2		
BS 102	English	CC-1A	5	5		
BS 103	Second Language	CC-2A	5	5		
BS 104	Chemistry of Biomolecules	DSC - 1A	4T+2P=6	4+1=5		
BS 105	Optional II	DSC - 2A	4T+2P=6	4+1=5		
BS 106	Optional III	DSC - 3A	4T+2P=6	4+1=5		
	TOTAL			27		
SEMESTER II						
BS 201	Gender Sensitization	AECC 2	2	2		
BS 202	English	CC-1B	5	5		
BS 203	Second Language	CC-2B	5	5		
BS 204	Chemistry of Nucleic acids	DSC -1B	4T+2P=6	4+1=5		
	and Biochemical					
	Techniques					
BS 205	Optional II	DSC -2B	4T+2P=6	4+1=5		
BS 206	Optional III	DSC -3B	4T+2P=6	4+1=5		
	TOTAL			27		
SECOND YEAR- SEMESTER III						
BS 301	SEC	SEC -1	2	2		
BS 302	English	CC-1C	5	5		
BS 303	Second Language	CC-2C	5	5		
BS 304	Bioenergetics, Biological	DSC- 1C	4T+2P=6	4+1=5		
	oxidation and Enzymology					
BS 305	Optional II	DSC- 2C	4T+2P=6	4+1=5		
BS 306	Optional III	DSC- 3C	4T+2P=6	4+1=5		
	TOTAL			27		
SEMESTE	CR IV	<del></del>		<del>_</del>		
BS 401	SEC	SEC – 2	2	2		
BS 402	English	CC-1D	5	5		
BS 403	Second Language	CC-2D	5	5		
BS 404	Intermediary Metabolism	DSC- 1D	4T+2P=6	4+1=5		
BS 405	Optional II	DSC- 2D	4T+2P=6	4+1=5		
BS 406	Optional III	DSC- 3D	4T+2P=6	4+1=5		
	TOTAL			27		

THIRD YEAR- SEMESTER V					
BS 501	SEC	SEC-3	2	2	
BS 502	Generic Elective	GE-1	2	2	
BS 503	Physiology and Clinical Biochemistry	DSC-1E	3T+2P=5	3+1=4	
BS 504	Optional II	DSC-2E	3T+2P=5	3+1=4	
BS 505	Optional III	DSC-3E	3T+2P=5	3+1=4	
BS 506	A- Molecular Biology	DSE-1E	3T+2P=5	3+1=4	
	B- Cell Biology and Genetics				
BS 507	Optional II A/B/C	DSE-2E	3T+2P=5	3+1=4	
BS 508	Optional III A/B/C	DSE-3E	3T+2P=5	3+1=4	
	TOTAL			28	
SEMESTER VI					
BS 601	SEC	SEC- 4	2	2	
BS 602	Generic Elective	GE-2	2	2	
BS 603	Nutrition and Immunology	DSC-1F	3T+2P=5	3+1=4	
BS 604	Optional II	DSC-2F	3T+2P=5	3+1=4	
BS 605	Optional III	DSC-3F	3T+2P=5	3+1=4	
BS 606	A- Microbiology and r-DNA technology	DSE-1F	3T+2P=5	3+1=4	
	B- Biotechnology				
BS 607	Optional II A/B/C	DSE-2F	3T+2P=5	3+1=4	
BS 608	Optional III A/B/C	DSE-3F	3T+2P=5	3+1=4	
	TOTAL			28	
	TOTAL CREDITS			164	

CC- Core Course

AECC- Ability Enhancement Compulsory Course DSC- Discipline Specific Core SEC- Skill Enhancement Course

DSE- Discipline Specific Elective

GE- Generic Elective

HPW – Hours per week

#### DSC - 1A

## Semester – I: Paper-BS 104 (Theory): Chemistry of Biomolecules (4 Credits; 4 Hr/week)

#### **Credit- I: Introduction**

- 1. Scope of biochemistry
- 2. Water as biological solvent
- 3. Weak acids and bases
- 4. pH, buffers,
- 5. Biological Buffers
- 6. Henderson- Hasselbalch equation (Simple numerical problems)
- 7. Stereo chemistry with reference to carbohydrates & amino acids.

#### Credit - II: Carbohydrates

- 1. Classification of carbohydrates
- 2. Mono saccharide straight chain and ring structures
- 3. Reactions of monosaccharides, mutarotation
- 4. Amino sugars and glycosides
- 5. Disaccharides, oligosaccharides& polysaccharides
- 6. Storage and structural polysaccharides, glycosaminoglycan's and
- 7. Bacterial cell wall polysaccharides.

#### **Credit - III: Lipids**

- 1. Classification of lipids
- 2. Essential fatty acids
- 3. Reactions & properties of lipids
- 4. General properties and structures of neutral fats, waxes, phospholipids, sphingolipids, cholesterol, glycolipids.
- 5. Prostaglandins and lipoproteins.
- 6. Bio membranes, behavior of amphipathic lipids in water, formation of micelles, bilayers, vesicles
- 7. Membrane composition and fluid mosaic model.

#### **Credit – IV: Amino acids & proteins**

- 1. Classification, structure, stereochemistry and chemical reactions of amino acids.
- 2. Titration curve of glycine &pk values.
- 3. Essential, nonessential amino acids and non-protein amino acids.
- 4. Peptide bond- Nature and conformation, Naturally occurring peptides –Glutathione, encephalin.
- 5. Outlines of protein classification, structural organization of proteins: primary, secondary, tertiary and quaternary structures (ex. hemoglobin & myoglobin).
- 6. General properties of proteins, denaturation and renaturation of proteins.
- 7. Determination of amino acid composition of proteins.

- 1. Lehninger's Principles of Biochemistry Nelson.D.L. and Cox.M.M., Freeman & Co.
- 2. Biochemistry Berg.J.M., Tymoczko.J.L. and Stryer.L., Freeman & Co.
- 3. Biochemistry Voet.D and Voet., J.G., John Wiley & Sons
- 4. Textbook of Biochemistry West.E.S.,Todd.W.R,Mason.H.S..and. Bruggen, J.T.V., Oxford & IBH Publishers.
- 5. Outlines of Biochemistry Conn.E.E., Stumpf.P.K., Bruening, G and Doi.R.H., John Wiley & Sons .
- 6. Harper's Illustrated Biochemistry Murray, R.K., Granner.D.K. & Rodwell, V.W., McGraw-Hill
- 7. Bichemistry-Lippincott's Illustrated Reviews. Champe, P.C. and Harvey, R. A. Lippincott
- 8. Fundamentals of Biochemistry Jain, J.L., Jain, S., Jain, N. S. Chand & Co.
- 9. Biochemistry Satyanarayana. U and Chakrapani. U, Books & Allied Pvt. Ltd.

#### DSC - 1A

## Semester – I: Paper-BS 104 (practical): Qualitative Analysis of Biomolecules (1 Credits; 2 Hr/week)

- 1. Laboratory general safety procedures
- 2. Preparation of standard solutions
- 3. Determination of pKa values of amino acids by titration
- 4. Preparation of buffers
- 5. Qualitative identification of Carbohydrates
- 6. Qualitative identification of Amino acids
- 7. Qualitative identification of Lipids

- 1. Experimental Biochemistry-A student companion-Beedu Sashidhar Rao and Vijay Deshpande.
- 2. Laboratory Manual in Biochemistry- Jayaraman, J. Wiley Eastern

## **Model paper: Theory**

**Duration 3 hours** Max. Marks 80 **Section - A (Short Answer Type) Answer all Questions**  $8 \times 4 = 32 \text{ Marks}$ 1. 2. **3.** 4. 5. 6. 7. 8. Section - B (Essay Answer Type) Answer all Questions  $4 \times 12 = 48$  Marks 9. (A). (OR) **(B) 10.** (A) (OR) **(B)** 11. (A) (OR) **(B)** 12. (A)

(OR)

**(B)** 

### **Model Paper Practicals (end of semester)**

Duration 3 hours Max. Marks 25

- 1. Write the Principles for the following experiments 5 Marks
- 2. Major Experiment 10 Marks
- 3. Minor Experiment 5 Marks
- 4. Viva-Voce and Record 5 Marks

#### **DSC - 1B**

### Semester – II: Paper-BS 204 (Theory) Chemistry of Nucleic Acids and Biochemical Techniques (4 Credits; 4 Hr/week)

#### **Credit - I: Composition of Nucleic acids**

- 1. Nature (functions) of nucleic acids.
- 2. Structure of purines and pyrimidines.
- 3. Nucleosides and Nucleotides
- 4. DNA & RNA.
- 5. Stability and formation of phosphodiester linkages
- 6. Effect of acids, alkali and nucleases and phosphodiester linkages
- 7. Photochemical and Spectral characteristics of Nucleic acid.

#### Credit - II: Structure of nucleic acids

- 1. Watson& Crick DNA double helix structure.
- 2. Introduction to circular DNA, supercoiling, helix to random coil transition,
- 3. denaturation of nucleic acids.
- 4. Hyper chromic effect
- 5. Tm values and their significance.
- 6. Reassociation kinetics, cot curves and their significance.
- 7. Different types of RNA and their biological functions.

#### Credit - III: Spectrophotometric and Centrifugation Techniques

- 1. Colorimetry and spectrophotometry.
- 2. Beer-Lamberts law and its limitations.
- 3. UV spectra
- 4. Visible spectra
- 5. Molar extinction coefficient.
- 6. Principle of fluorimetry
- 7. Principle and applications of Centrifugation technique in biology

#### **Credit – IV: Chromatography techniques**

- 1. Principle in chromatographic technique.
- 2. Application of chromatographic technique in paper chromatography
- 3. TLC
- 4. Gel filtration (molecular sieve)
- 5. Ion exchange Chromatography
- 6. Affinity chromatography.

- 1. Biochemistry Voet.D and Voet., J.G., John Wiley & Sons
- 2. Textbook of Biochemistry West.E.S.,Todd.W.R,Mason.H.S..and. Bruggen, J.T.V., Oxford & IBH Publishers.
- 3. Outlines of Biochemistry Conn.E.E., Stumpf.P.K., Bruening, G and Doi.R.H., John Wiley & Sons .
- 4. Principles and Techniques of Practical Biochemistry-Wilson, K. and Walker, J. Cambridge Press.
- 5. The Tools of Biochemistry- Cooper, T. G. John Wiley & Sons Press.
- 6. Physical Biochemistry-Friefelder, D. W.H. Freeman Press.
- 7. Analytical Biochemistry Holme.D.J. and Peck.H., Longman.
- 8. Biophysical Chemistry: Principle and techniques- Upadhyay A, Upadhyay K and Nath. N. Himalaya Publishing House.
- 9. Experimental Biochemistry- Clark Jr. J.M and Switzer, R. L. Freeman & Co..

#### **DSC - 1B**

# Semester – II: Paper-BS 204 (practical): Quantitative Analysis of Biomolecules (1 Credits; 2 Hr/week)

- 1. Amino acid Estimation by Ninhydrin method
- 2. Protein Estimation by Folin's Method
- 3. Total Sugar Estimation by Anthrone Method
- 4. Total Reducing Sugar Estimation by Dinitrosalicylate
- 5. Estimation of Keto sugar by Roe's resorcinol Method

- 1. Experimental Biochemistry-A student companion-Beedu Sashidhar Rao and Vijay Deshpande.
- 2. Laboratory Manual in Biochemistry- Jayaraman, J. Wiley Eastern

## **Model paper: Theory**

**Duration 3 hours** Max. Marks 80 **Section - A (Short Answer Type) Answer all Questions**  $8 \times 4 = 32 \text{ Marks}$ 1. 2. **3.** 4. 5. 6. 7. 8. Section - B (Essay Answer Type) Answer all Questions  $4 \times 12 = 48$  Marks 9. (A). (OR) **(B)** 10. (A) (OR) **(B)** 11. (A) (OR) **(B)** 

12. (A)

**(B)** 

(OR)

## **Model Paper Practicals (end of semester)**

Duration 3 hours Max. Marks 25

1. Write the Principles for the following experiments 5 Marks

2. Major Experiment 10 Marks

3. Minor Experiment 5 Marks

4. Viva-Voce and Record 5 Marks

#### DSC - 1C

# Semester – III: Paper-BS 304 (Theory): BIOENERGETICS, BIOLOGICAL OXIDATIONS AND ENZYMOLOGY (4 Credits; 4 Hr/week)

#### **Credit- I: Bioenergetics**

- 1. Energy transformations in the living system
- 2. Free energy, Enthalpy and Entropy concepts.
- 3. Exergonic and endergonic reactions.
- 4. High energy compounds.
- 5. Phosphate group transfer potential.
- 6. Substrate level phosphorylation.
- 7. Cytochromes-structure, types and their functions

#### **Credit – II: Biological Oxidations**

- 1. Biological oxidations: Definition, enzymes involved- oxidases, dehydrogenases and oxygenases.
- 2. Redox reactions. Redox couplers. Reduction potential ( $\varepsilon$ ,  $\varepsilon_0$ ,  $\varepsilon'_0$ ). Standard reduction potential ( $\varepsilon'_0$ ) of some biochemically important half reactions.
- 3. Ultrastructure of mitochondria, Electron transport chain and carriers involved.
- 4. Oxidative phosphorylation, theories of oxidative phosphorylation-Mitchell's chemiosmotic theory.  $F_0$   $F_1$  ATPase, Inhibitors of respiratory chain and oxidative phosphorylation, uncouplers.
- 5. Formation of reactive oxygen species and their disposal through enzymatic reactions.
- 6. Ultrastructure of chloroplast
- 7. Cyclic and non-cyclic photophosphorylation.

#### **Credit- III: Introduction to Enzymology**

- 1. Introduction to biocatalysis, differences between chemical and biological catalysis.
- 2. Nomenclature and classification of enzymes.
- 3. Definition of holo-enzyme, apo-enzyme, coenzyme, cofactor, Fundamentals of enzyme assay, enzyme units.
- 4. Methods of Enzyme purification
- 5. Enzyme specificity. Active site.
- 6. Principles of energy of activation, transition state.
- 7. Interaction between enzyme and substrate- lock and key, induced fit models.

### Credit – IV: Enzyme Kinetics and Enzyme action

- 1. Rate of a Reaction Law of Mass action, Factors affecting the catalysis- substrate concentration, *pH*, temperature, Time, Enzyme concentration and Product concentration
- 2. Michaelis Menten equation for single substrate reaction, significance of  $K_{\rm M}$  and  $V_{\rm max}$ .
- 3. Enzyme inhibition- irreversible and reversible, types of reversible inhibitions- competitive and non-competitive.
- 4. Outline of mechanism of enzyme action- acid-base catalysis, covalent catalysis, electrostatic catalysis, and metal ion catalysis.
- 5. Regulation of enzyme activity- allosterism and cooperatitvity, ATCase as an allosteric enzyme, covalent modulation- covalent phosphorylation of phosphorylase
- 6. Zymogen activation- activation of trypsinogen and chymotrypsinogen.
- 7. Isoenzymes (LDH) and Multienzyme complexes (PDH). Ribozyme.

- 1. Lehninger's Principles of Biochemistry Nelson.D.L. and Cox.M.M., Freeman & Co.
- 2. Biochemistry Berg.J.M., Tymoczko.J.L. and Stryer.L., Freeman & Co.
- 3. Biochemistry Voet.D and Voet., J.G., John Wiley & Sons
- 4. Textbook of Biochemistry West.E.S., Todd.W.R, Mason.H.S..and. Bruggen, J.T.V., Oxford & IBH Publishers.
- 5. Outlines of Biochemistry Conn.E.E., Stumpf.P.K., Bruening, G and Doi.R.H., John Wiley & Sons .
- 6. Harper's Illustrated Biochemistry Murray, R.K., Granner, D.K. & Rodwell, V.W., McGraw-Hill
- 7. Bichemistry-Lippincott's Illustrated Reviews. Champe, P.C. and Harvey, R. A. Lippincott
- 8. Fundamentals of Biochemistry Jain, J.L., Jain, S., Jain, N. S. Chand & Co.
- 9. Biochemistry Satyanarayana. U and Chakrapani. U, Books & Allied Pvt. Ltd.
- 10. Fundamentals of Enzymology Price.N.C.and Stevens.L., Oxford University Press.
- 11. Understanding Enzymes Palmer.T., Ellis Harwood.
- 12. Enzymes Biochemistry, Biotechnology, Clinical Chemistry Palmer.T., Affiliated East-West Press

### DSC – 1C Semester – III: Paper - BS 304 (Practicals): ENZYMOLOGY (1 Credits; 2 Hr/week)

- 1. Assay of salivary  $\alpha$ -amylase
- 2. Assay of  $\beta$ -amylase from sweet potatoes
- 3. Assay of urease
- 4. Assay of catalase
- 5. Assay of phosphatase
- 6. Determination of optimum temperature and pH for amylase
- 7. Determination of optimum pH for phosphatase
- 8. Effect of Substrate concentration of amylase activity

- 1. Experimental Biochemistry-A student companion-Beedu Sashidhar Rao and Vijay Deshpande.
- 2. Laboratory Manual in Biochemistry- Jayaraman, J. Wiley Eastern
- 3. Enzyme Assays- A practical Approach: Eisenthal, R and Dawson, M.l., IRL Press.
- 4. Biochemical Methods- Sadasivam, S and Manickyam, A. New Age International Publishers.

### **Model paper: Theory**

Duration 3 hours Max. Marks 80 **Section - A (Short Answer Type) Answer all Questions**  $8 \times 4 = 32 \text{ Marks}$ 1. 2. **3.** 4. 5. 6. 7. 8. Section - B (Essay Answer Type) Answer all Questions  $4 \times 12 = 48$  Marks 9. (A). (OR) **(B) 10.** (A) (OR) **(B)** 11. (A) (OR) **(B)** 

12. (A)

**(B)** 

(OR)

## **Model Paper Practicals (end of semester)**

Duration 3 hours Max. Marks 25

- 1. Write the Principles for the following experiments 5 Marks
- 2. Major Experiment 10 Marks
- 3. Minor Experiment 5 Marks
- 4. Viva-Voce and Record 5 Marks

#### **DSC-1D**

## Semester – IV: Paper-BS 404 (Theory): INTERMEDIARY METABOLISM (4 Credits; 4 Hr/week)

#### Credit-I: Amino acid Metabolism

- 1. General reactions of amino acid metabolism-transamination, decarboxylation and deamination
- 2. Urea cycle and regulation
- 3. Catabolism of carbon skeleton of amino acids- glycogenic and ketogenic amino acids.
- 4. Metabolism of glycine, serine, aspartic acid, methionine, phenylalanine and leucine.
- 5. Biosynthesis of creatine.
- 6. Inborn errors of aromatic amino acids
- 7. Inborn errors of branched chain amino acid metabolism.

#### Credit- II: Carbohydrate Metabolism

- 1. Concept of anabolism and catabolism.
- 2. Glycolytic pathway, energy yield. Fate of pyruvate- formation of lactate and ethanol, Pasteur effect.
- 3. Citric acid cycle, regulation, energy yield, amphipathic role. Anaplerotic reactions.
- 4. Glycogenolysis and glycogenesis.
- 5. Pentose phosphate pathway.
- 6. Gluconeogenesis.
- 7. Photosytnthesis- Light and Dark reactions, Calvin cycle and C<sub>4</sub> Pathway, CAM Pathway

#### **Credit – III: Lipid Metabolism**

- 1. Catabolism of fatty acids (β- oxidation) with even and odd number of carbon atoms
- 2. Ketogenesis
- 3. *de novo* synthesis of fatty acids
- 4. Elongation of fatty acids in mitochondria and microsomes
- 5. Biosynthesis and degradation of triacylglycerol
- 6. Biosynthesis of lecithin.
- 7. Biosynthesis of cholesterol.

#### Credit - IV: Nucleic acid Metabolism

- 1. Biosynthesis of purine and pyrimidine nucleotides, *de novo* and salvage pathways.
- 2. Regulation of purine and pyrimidine nucleotides
- 3. Catabolism of purines and pyrimidines.
- 4. Biosynthesis of deoxyribonucleotides- ribonucleotide reductase and thymidylate synthase and their significance.
- 5. Disorders of nucleotide metabolism- Gout, Lesch- Nyhan syndrome.
- 6. Biosynthesis of heme
- 7. Degradation of heme

- 1. Lehninger's Principles of Biochemistry Nelson.D.L. and Cox.M.M., Freeman & Co.
- 2. Biochemistry Berg.J.M., Tymoczko.J.L. and Stryer.L., Freeman & Co.
- 3. Biochemistry Voet.D and Voet., J.G., John Wiley & Sons .
- 4. Textbook of Biochemistry West.E.S., Todd.W.R, Mason.H.S..and. Bruggen, J.T.V., Oxford & IBH Publishers.
- 5. Principles of Biochemistry: General Aspects-Smith, E. L., Hill, R.L. Lehman, I. R. Lefkowitz, R.J. Handler, P., and White, A. McGraw-Hill
- 6. Outlines of Biochemistry Conn.E.E., Stumpf.P.K., Bruening, G and Doi.R.H., John Wiley & Sons .
- 7. Harper's Illustrated Biochemistry Murray, R.K., Granner.D.K. & Rodwell, V.W., McGraw-Hill 8. Bichemistry-Lippincott's Illustrated Reviews. Champe, P.C. and Harvey, R. A. Lippincott
- 8. Fundamentals of Biochemistry Jain, J.L., Jain, S., Jain, N. S. Chand & Co.
- 9. Biochemistry Satyanarayana. U and Chakrapani. U, Books & Allied Pvt. Ltd.
- 10. Biochemistry Rama Rao. A and Ratna Kumari. D, Kalyani Publishers.
- 11. Biochemistry- The Molecular Basis of Life McKee. T and McKee, J. R, McGraw-Hill.

### DSC – 1 D Paper-BS 404 (Practicals): BIOCHEMICAL PREPARATIONS AND SEPARATIONS (1 Credits; 2 Hr/week)

- 1. Absorption maxima of colored substances- p-Nitrophenol, Methyl orange and KMnO<sub>4</sub>.
- 2. Absorption spectra of Amino acid Tyrosine; protein-BSA, nucleic acids- Calf thymus DNA.
- 3. Isolation of egg albumin from egg white.
- 4. Isolation of cholesterol from egg yolk.
- 5. Isolation of starch from potatoes.
- 6. Isolation of casein from milk.
- 7. Separation of amino acids by Paper chromatography
- 8. Separation of Plant pigments by TLC

- 1. Experimental Biochemistry-A student companion-Beedu Sashidhar Rao and Vijay Deshpande.
- 2. Laboratory Manual in Biochemistry- Jayaraman, J. Wiley Eastern

## **Model paper: Theory**

Max. Marks 80

Duration 3 hours

	Duration 3 not		Max. Marks 80		
	Section - A (Short Answer Type)				
		<b>Answer all Questions</b>	$8 \times 4 = 32 \text{ Marks}$		
1.					
2.					
3.					
4.					
5.					
6.					
7.					
8.					
		Section - B (Essay A	Section - B (Essay Answer Type)		
		Answer all Questions 4	x 12 = 48 Marks		
		inswer an Questions	A 12 — 10 IVILIAS		
9. (A).					
)• (I <b>1</b> )•	(OR)				
<b>(B)</b>					
10 (1)					
<b>10.</b> (A)	(OR)				
<b>(B)</b>	(OK)				
11. (A)					
( <b>T</b> )	(OR)				
<b>(B)</b>					
12. (A)					
	(OR)				
<b>(B)</b>					

## **Model Paper Practicals (end of semester)**

Duration 3 hours Max. Marks 25

- 1. Write the Principles for the following experiments 5 Marks
- 2. Major Experiment 10 Marks
- 3. Minor Experiment 5 Marks
- 4. Viva-Voce and Record 5 Marks

#### DSC - 1E

## Semester – V: Paper-BS 503 (Theory): Physiology and Clinical Biochemistry (3 Credits; 3 Hr/week)

#### **Credit-1: Physiology**

- 1. Digestion and absorption of carbohydrates, lipids and proteins
- 2. Composition of blood and coagulation of blood
- 3. Hemoglobin and transport of gases in blood (oxygen and CO<sub>2</sub>)
- 4. Heart- structure of the heart
- 5. Cardiac cycle, cardiac factors controlling blood pressure.
- 6. Muscle- kinds of muscles, structure of myofibril, organization of contractile proteins and mechanism of muscle contraction.
- 7. Structure of Neuron and propagation of nerve impulse

#### **Credit-2: Endocrinology**

- 1. Endocrinology- organization of endocrine system. Classification of hormones.
- 2. Mechanism of hormonal action- signal transduction pathways for adrenaline, glucocorticoids and insulin.
- 3. Chemistry, physiological role and disorders of hormones of Pituitary, Hypothalamus and Thyroid
- 4. Chemistry, physiological role and disorders of hormones of Pancreas
- 5. Chemistry, physiological role and disorders of hormones of Parathyroid
- 6. Chemistry, physiological role and disorders of hormones of Gonads, Placenta and Adrenals
- 7. Introduction of gastrointestinal hormones.

#### **Credit-3: Organs and Organ Function tests**

- 1. Structure and functions of the liver.
- 2. Liver function tests- conjugated and total bilurubin in serum, albumin: globulin ratio, hippuric acid and bromsulphthalein tests. Serum enzymes in liver diseases- SGPT, GGT and alkaline phosphatase.
- 3. Kidneys-structure of nephron and Mechanism of urine formation
- 4. Normal and abnormal constituents of urine.
- 5. Biological buffers. Role of kidneys in maintaining acid-base and electrolyte balance in the body.
- 6. Renal function tests- creatinine and urea clearance tests, phenol red test.
- 7. Biochemical tests for the diagnosis of heart diseases- HDL/LDL cholesterol, SGOT, LDH, CK, Creactive protein, cardiac troponins.

- 1. Textbook of Biochemistry and Human Biology Talwar, G.P. and Srivastava. L.M., Printice Hall of India
- 2. Review of Medical Physiology-Ganong. McGraw-Hill.
- 3. Human Physiology Chatterjee.C.C, Medical Allied Agency
- 4. Textbook of Medical Physiology Guyton.A.G and Hall.J.E., Saunders
- 5. William's Textbook of Endocrinology Larsen, R. P. Korenberg, H. N. Melmed, S. and Polensky, K. S. Saunders
- 6. Mammalian Biochemistry- White, A. Handler, P. and Smith, E. L. McGraw-Hill.

- 7. Textbook of Human Nutrition-Bamji, Pralhad Rao and Reddy V. Oxford & IBH Publishers.
- 8. Foods: Facts & Principle- Shakuntala and Shadaksharaswamy. Wiley Ester Press.
- 9. Essentials of Food and Nutrition Swaminathan.M. Bangalore Press.
- 10. Human Nutrition and Dietetics. Davidson, S. and Passmore, J. R. ELBS.
- 11. A Textbook of Biochemistry: Molecular and Clinical Aspects. Nagini, S. Scitech Publishers.
- 12. *Tietz* Fundamentals of Clinical Chemistry- Burtis, A. A. and Ashwood, E. R. Saundersimprint Elsevier Pub.
- 13. Textbook of Biochemistry with Clinical Correlations Devlin.T.M., Wiley Liss
- 14. Textbook of Medical Biochemistry Chatterjea.M.N. and Shinde.R, Jaypee Brothers Medical Publishers
- 15. Textbook of Medical Biochemistry- Ramakrishnan, S., Prasannan, K. G. and Rajan, R. Orient Longman

#### **DSC - 1E**

## Semester – V: Paper - BS 503 (Practicals): Physiology and Clinical Biochemistry (1 Credits; 2 Hr/week)

- 1. Estimation of hemoglobin in blood.
- 2. Total count RBC and WBC. Differential count.
- 3. Urine analysis for albumin, sugars and ketone bodies.
- 4. Estimation of urinary creatinine.
- 5. Estimation of blood urea.
- 6. Estimation of serum total cholesterol.
- 7. Determination of serum alkaline phosphatase activity.
- 8. Determination of SGOT and SGPT activity

- 1. Experimental Biochemistry-A student companion-Beedu Sashidhar Rao and Vijay Deshpande.
- 2. Laboratory Manual in Biochemistry- Jayaraman, J. Wiley Eastern
- 3. Biochemical Methods- Sadasivam, S and Manickyam, A. New Age International Publishers

#### DSE - 1E

## Semester – V: Paper-BS 506 A (Theory): Molecular Biology (3 Credits; 3 Hr/week)

#### **Credit- I: DNA Replication**

- 1. Organization of genome in prokaryotes and eukaryotes.
- 2. Experimental evidences to prove nucleic acids as genetic material.
- 3. Nature and structure of the gene.
- 4. DNA replication- models of replication, Meselson-Stahl's experimental proof for semi-conservative model.
- 5. DNA polymerases I, II and III of *E.coli*, helicase, topoisomerases, primase, ligase.
- 6. Bidirectional replication model. Okazaki fragments, leading and lagging strands of DNA synthesis.
- 7. Inhibitors of DNA replication.

#### **Credit- II: Transcription**

- 1. Transcription RNA synthesis, RNA polymerases of prokaryotes.
- 2. Promoters, Initiation- sigma factors and their recognition sites.
- 3. Elongation-role of core enzyme.
- 4. Termination- rho dependent and rho independent. RNA polymerase I, II and III of eukaryotes.
- 5. Transcriptional events in eukaryotic m-RNA synthesis
- 6. Post-transcriptional modifications of eukaryotic m-RNA. I
- 7. Inhibitors of RNA synthesis.

#### Credit- III: Translation and Regulation of Gene Expression

- 1. Introduction to protein synthesis- Genetic code, structure of t-RNA
- 2. Deciphering of genetic code, Nirenberg's and Khorana's experiments, wobble hypothesis, degeneracy of genetic code.
- 3. Protein synthesis- activation of amino acids (aminoacyl t-RNA synthetases).
- 4. Ribosome structure. Initiation, elongation and termination of protein synthesis.
- 5. Post- translational modifications- signal hypothesis.
- 6. Inhibitors of protein synthesis.
- 7. Regulation of prokaryotic gene expression- induction and repression. Lac operon, catabolite repression. Tryptophan operon and attenuation.

- 1. Molecular Biology of Cell- Alberts, B. Bray, D. Lewis, J. Raff, M. Roberts, K. and Watson, J. D. Garland Publishing.
- 2. Recombinant DNA and Biotechnology: A Guide for teachers- Helen and Massey. ASM Press.
- 3. Genes VIII Lewin. B, Oxford University Press.
- 4. Molecular Biology- Freifelder. D. Naroasa Pub. House
- 5. Molecular Biology of the Gene- Watson. J.D., Baker, T.A, Bell, S.P., Gann.A, Levine, M. and Losick.R, Pearson Education.
- 6. Molecular Biotechnology- Glick, B. R. and Pasternak, J. J. ASM Press
- 7. Principles of Gene Manipulation: An Introduction to GE- Old, R. V. and Primrose, S. B. Blackwell Sci. Pub.
- 8. Molecular Cell Biology- Lodish, H., Berk, A., Matsudaira, P., Kaiser, C. A., Krieger, M. Scott M. P., Zipursky, S. L. and Darnell, J. Freeman & Co.

## DSE – 1 E Semester – V: Paper - BS 506 A (Practicals) : Molecular Biology (1 Credits; 2 Hr/week)

- 1. Isolation of DNA from onion/liver/coconut endosperm.
- 2. Isolation of plasmids
- 3. Isolation of RNA
- 4. Determination of purity of nucleic acids by UV-spectrophotometric method.
- 5. Estimation of DNA by diphenylamine method.
- 6. Estimation of RNA by orcinol method.
- 7. Electrophoresis of nucleic acids and visualization by methylene blue staining.
- 8. Restriction mapping:  $\lambda$  DNA with any two restriction enzymes.

- 1. Experimental Biochemistry-A student companion-Beedu Sashidhar Rao and Vijay Deshpande.
- 2. Laboratory Manual in Biochemistry- Jayaraman, J. Wiley Eastern
- 3. Biochemical Methods- Sadasivam, S and Manickyam, A. New Age International Publishers

#### DSE - 1E

## Semester – V: Paper-BS 506 B (Theory): Cell Biology and Genetics (3 Credits; 3 Hr/week)

### Credit - I: Cell Biology

- 1. Cells as basic units of living organisms
- 2. Composition & functions of cell organelles
- 3. Cytoskeleton- Microfilaments, Microtubules & Intermediate filaments
- 4. Ultra structure of prokaryotic cell and eukaryotic cells
- 5. Chromosome organization in Prokaryotes and Eukaryotes and structure of chromosomes (Polytene and Lamp Brush)
- 6. Mitosis and meiosis Significance of mitosis and meiosis
- 7. Cell Cycle and cell death

#### **Credit – II: Genetics**

- 1. Basic concepts of Mendel's experiments Law of segregation and Law of Independent assortment
- 2. Partial or incomplete dominance and Co-dominance
- 3. Non-Mendelian inheritance: Extra chromosomal inheritance (paramecium & drosophila).
- 4. Maternal inheritance (Coiling in snails, Laber's hereditary optic neuropathy).
- 5. Linkage and recombination
- 6. Polygenic inheritance (Introduction to quantitative traits).
- 7. Sex linked inheritance. X-linked recessive inheritance (colour blindness & Hemophilia)

#### **Credit – III: Mutations and Mutagens**

- 1. Mutations (spontaneous / induced, somatic / germinal, forward / reverse, transition / transversions)
- 2. Mutations (Silent, missense, nonsense, and frame shift mutations, conditional, leaky)
- 3. Detection, selection & isolation of microbial mutants
- 4. Estimation of mutation rates
- 5. Reversion and suppression of mutations
- 6. Mutagens physical, chemical
- 7. Transposon mutagenesis, site-directed mutagenesis

- 1. Principles of Genetics by Eldon John Gardner, Michael J. Simmons, D. Peter Snustad; John Wiley
- 2. Modern Genetic Analysis Anthony JF Griffiths, William M Gilbert, Jeffrey H Miller, and Richard C Lewontin. Pub. W. H. Freeman
- 3. Lewin B. (Ed.) (1996) Genes, VII edition, John Wiley and Sons, New York.
- 4. Cell and Molecular Biology, De Robertis and De Robertis , Lippincott & Wilkins
- 5. Cell Biology by C. B. Pawar
- 6. Principles of Genetics by R.H. Tamarin McGrawhill
- 7. Theory & problems in Genetics by Stansfield, Schaum out line series McGrawhill

#### **DSE - 1 E**

## Semester – V: Paper-BS 506 B (Practicals) : Cell Biology and Genetics (1 Credits; 2 Hr/week)

- 1. Preparation of different stages of Mitosis
- 2. Preparation of different stages of Meiosis
- 3. Types of chromosomes
- 4. Karyotyping
- 5. Problems on Monohybrid cross
- 6. Problems on dihybrid ratio in *Drosophila*/maize
- 7. Problems on Linkage and Recombination
- 8. Studies on Sex linked inheritance and X-linked recessive inheritance

#### References

1. Essential practical handbook of Cell Biology & Genetics, Biometry and Microbiology: A Laboratory Manual by Debarati Das, Academic Publishers

## **Model paper: Theory**

Duration 3 hours Max. Marks 75 **Section - A (Short Answer Type) Answer all Questions**  $6 \times 5 = 30 \text{ Marks}$ 1. 2. **3.** 4. 5. 6. Section - B (Essay Answer Type) Answer all Questions  $3 \times 15 = 45$  Marks 7. (A). (OR) **(B)** 8. (A) (OR) **(B)** 

9. (A)

**(B)** 

(OR)

### **Model Paper Practicals (end of semester)**

Duration 3 hours Max. Marks 25

- 1. Write the Principles for the following experiments 5 Marks
- 2. Major Experiment 10 Marks
- 3. Minor Experiment 5 Marks
- 4. Viva-Voce and Record 5 Marks

#### **DSC - 1 F**

## Semester – VI: Paper-BS 603 (Theory): Nutrition and Immunology (3 Credits; 3 Hr/week)

#### **Credit – I: Nutrition**

- 1. Balanced diet. Calorific values of foods and their determination by bomb calorimeter.
- 2. BMR and factors affecting BMR. Specific dynamic action of foods.
- 3. Energy requirements and recommended dietary allowance (RDA) for children, adults, pregnant and lactating women.
- 4. Sources of complete and incomplete proteins. Biological value of proteins. Role of essential fatty acids in human nutrition.
- 5. Malnutrition- Kwashiorkar, Marasmus and PEM.
- 6. Vitamins sources, structure, biochemical roles, deficiency disorders of water and fat soluble vitamins; Bulk and trace elements Ca, Mg, Fe, I, Cu, Mo, Zn, Se and F.
- 7. Nutraceuticals; Obesity and starvation.

#### **Credit – II: Immunology**

- 1. Organization of immune system.
- 2. Organs and cells of immune system.
- 3. Innate and acquired immunity.
- 4. Cell mediated and humoral immunity (T- and B- cells).
- 5. Classification of immunoglobulins, structure of IgG. Theories of antibody formation- clonal selection theory.
- 6. Epitopes / antigenic determinants. Concept of haptens. Adjuvants.
- 7. Monoclonal antibodies and their applications

#### Credit – III: Immunotechnology

- 1. Antigen-antibody reactions- agglutination, immunoprecipitation, immunodiffusion.
- 2. Blood group antigens.
- 3. Immunodiagnostics-RIA, ELISA.
- 4. Vaccines and their classification. Traditional vaccines-live and attenuated, toxoids.
- 5. Modern vaccines- recombinant and peptide vaccines.
- 6. Outlines of hypersensitivity reactions.
- 7. Fundamentals of graft rejection and MHC proteins.

- 1. Essentials of Food and Nutrition –Swaminathan M. Bangalore Press
- 2. Immunology. Tizard, I. R. Thomson Press.
- 3. Kuby Immunology Kindt.T.J., Goldsby.R.A. and Osborne.B.A., Freeman & Co.
- 4. Roitt's Essential Immunology Roitt.I.M. and Delves.P.J., Blackwell Science.

### DSC - 1F

## Semester – VI: Paper-BS 603 (Practicals): Nutrition and Immunology (1 Credits; 2 Hr/week)

- 1. Estimation of calcium by titrimetry
- 2. Estimation of iron in apple juice by phenanthroline method.
- 3. Estimation of vitamin C by 2, 6 -dichlorophenol indophenol method.
- 4. Isolation of total lipids by gravimetric method.
- 5. Determination of iodine value of an oil.
- 6. Determination of acid value of an oil.
- 7. Agglutination: ABO and D Ag typing
- 8. ODD and ELISA sandwich ELISA

- 1. Experimental Biochemistry-A student companion-Beedu Sashidhar Rao and Vijay Deshpande.
- 2. Laboratory Manual in Biochemistry- Jayaraman, J. Wiley Eastern
- 3. Biochemical Methods- Sadasivam, S and Manickyam, A. New Age International Publishers

#### DSE - 1F

## Semester – VI: Paper-BS 606 A (Theory): Microbiology and r-DNA Technology (3 Credits; 3 Hr/week)

#### Credit – I: Microbiology

- 1. Introduction to brief history of microbiology. Classification of microorganisms- prokaryotic and eukaryotic microorganisms.
- 2. Isolation and cultivation of bacteria. Selective media and enriched media.
- 3. Bacterial growth curve and kinetics of growth. Batch, continuous and synchronous cultures.
- 4. Gram's staining- Gram positive and Gram negative bacteria, motility and sporulation.
- 5. Industrial uses of Aspergillus niger, yeast and Spirulina.
- 6. Structure and composition of viruses. One-step growth and determination of plaque forming units (PFU).
- 7. Isolation and cultivation of bacterial plaques. Lytic and lysogenic life cycle of  $\lambda$  phage. TMV, Retro viruses- HIV. Prions and Mycoplasma.

#### Credit - II: r-DNA technology I

- 1. Outlines of cloning strategies.
- 2. DNA sequencing- Maxam Gilbert and Sanger's methods.
- 3. Tools of r-DNA technology: Enzymes- Restriction endonucleases and ligases
- 4. Restriction mapping.
- 5. Cloning vectors- Plasmids, Cosmids, and  $\lambda$  phages
- 6. Hosts- E.coli
- 7. Molecular markers RFLP, AFLP and RAPD

#### Credit – III: r-DNA technology II

- 1. Construction of c-DNA libraries.
- 2. Polymerase chain reaction- principle and applications.
- 3. Outlines of blotting techniques-Southern, Northern and Western.
- 4. Applications of gene cloning- production of insulin
- 5. Production of human growth hormone
- 6. Production of Bt cotton
- Edible vaccines.

- 1. Textbook of Microbiology Ananthanarayan, R and Jayaram Paniker, C.K., Orient Longman.
- 2. Microbiology Prescott.L.M., Harley.J.P. & Klein.D.A, McGraw-Hill.
- 3. Microbiology Pelczar Jr, M.J., Chan.E.C.S. and Krieg.N.R., Tata McGraw-Hill.
- 4. Textbook of Microbiology- Dubey, R. C. and Maheshwari, D. K. S. Chand & Co.
- 5. Principles of Gene Manipulation: An introduction to GE Old, R. and Primrose, S.B. Blackwell Sci. Pub
- 6. Molecular Biotechnology Glick, BR and Paternak, JJ. Publish ASM Press

#### DSE - 1F

## Semester – VI: Paper-BS 606 A (Practicals): Microbiology and r-DNA Technology (3 Credits; 3 Hr/week)

- 1. Preparation of culture media and sterilization methods.
- 2. Isolation of pure cultures: (i) Streak plate method (ii) Serial dilution method.
- 3. Gram staining.
- 4. Motility of bacteria by hanging drop method.
- 5. Bacterial growth curve.
- 6. Antibiotic sensitivity by paper disc method.
- 7. Gene cloning (Demonstration only)
- 8. Preparation and transformation of competent cells

- 1. Molecular Cloning (Lab manual) by Maniatis T, Fritsch EF, Sambrook J, Volume –I, CSH
- 2. Microbiology A Laboratory manual by Cappuccino and Sherman, Pearson Publications LPE.
- 3. Experiments in Microbiology, Plant Pathology and Biotechnology by Aneja A. R., New Age Publications

## DSE – 1 F Semester – VI: Paper-BS 606 B (Theory): Biotechnology (3 Credits; 3 Hr/week)

#### **Credit – I: Plant Biotechnology**

- 1. Plant tissue culture and its applications
- 2. Plants as bioreactors and valuable chemical factories (production of bioactive compounds)
- 3. Crop improvement, Production of herbicide and insect resistant plants
- 4. Plant metabolic engineering
- 5. Genetic engineering for quality improvement of Protein, lipids, carbohydrates, vitamins & mineral nutrients
- 6. Marker-assisted selection of qualitative and quantitative traits.
- 7. Genetically modified crops Golden rice, soybeans, Bt cotton, tobacco, potato, papaya, jatropha, Arabidopsis

#### Credit – II: Animal and Microbial Biotechnology

- 1. Animal cell cultures as bioreactors
- 2. Usage of animal cell culture for in vitro drug testing
- 3. Molecular pharming; Production of vaccines, pharmaceutical proteins, recombinant hemoglobin and blood substituents
- 4. Microbes as biocontrol agents
- 5. Overview of Microbial insecticides (Baculoviruses, *Bacillus thurinigiensis* and *Bacillus sphaericus*)
- 6. Bioremediation, Biodegradation of cellulose and lignocellulose, biosurfactants and bioemulsifiers
- 7. Microbial ore leaching and production of microbial fuels (hydrogen, methane)

#### Credit – III: Environmental Biotechnology

- 1. Renewable and Non-renewable energy sources
- 2. Strategies involved in Municipal solid waste treatment
- 3. Treatment of industrial and domestic effluent, (aerobic and anaerobic)
- 4. Biomaterials as an alternative to non-degradable materials
- 5. Microorganisms for Heavy Metal Accumulation
- 6. Biosorption
- 7. Heavy metal tolerance (including mechanism) and its impact on environment

- 1. Introduction to Biotechnology, William J. Thieman, Michael A. Palladino, Benjamin Cummings Publ
- 2. Biotechnology- Arora, Himalaya pub. House
- 3. Introduction to Environmental Biotechnology by A. K. Chatterji, PHI Learning Pvt. Ltd.
- 4. Animal Cells as Bioreactors By Terence Gartoright, Cambridge Univ Press
- 5. Text Book of Biotechnology By H.K. Das (Wiley Publications)
- 6. Introduction to Plant Tissue Culture By M.K. Razdan (Oxford and IBH Publishing Company, New Delhi)
- 7. Industrial Microbiology by L.E. Casida

## DSE – 1 F Semester – VI: Paper-BS 606 B (Practicals): Biotechnology (1 Credits; 2 Hr/week)

- 1. Preparation of MS medium and initiation of callus
- 2. Micropropagation of plants
- 3. Preparation of animal cell culture media, Cell disaggregation and cell counting
- 4. Isolation of microbes from environment (soil, water, skin bread, milk)
- 5. Microbial degradation of organic matter
- 6. Efficacy testing for biofertilizers (nodulation test for rhizobia) and biopesticides
- 7. Municipal solid waste treatment and Waste water treatment
- 8. Production of hydrogen and methane

- 1. Microbial Biotechnology A Laboratory Manual for bacterial systems by Das, Surajit, Dash, Hirak Ranjan, Springer-Verlag
- 2. Plant Tissue Culture by Kalyan Kumar De
- 3. Biogas Technology by b.T. Nijaguna
- 4. Biotechnology procedures and experiments handbook by S. Harisha, Infinity Science Press LLC.

## **Model paper: Theory**

Duration 3 hours Max. Marks 75 **Section - A (Short Answer Type) Answer all Questions**  $6 \times 5 = 30 \text{ Marks}$ 1. 2. **3.** 4. 5. 6. Section - B (Essay Answer Type) Answer all Questions  $3 \times 15 = 45$  Marks 7. (A). (OR) **(B)** 8. (A) (OR) **(B)** 

9. (A)

**(B)** 

(OR)

## **Model Paper Practicals (end of semester)**

Duration 3 hours Max. Marks 25

- 1. Write the Principles for the following experiments 5 Marks
- 2. Major Experiment 10 Marks
- 3. Minor Experiment 5 Marks
- 4. Viva-Voce and Record 5 Marks